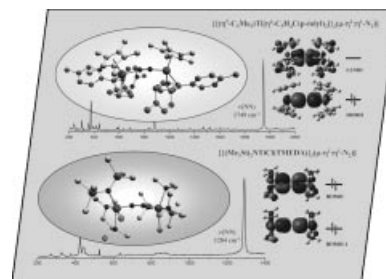


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COVER PICTURE

The cover picture shows that Raman spectroscopy can reveal drastically different N–N stretching frequencies in dinuclear, dinitrogen-bridged titanium complexes. In the article by F. Tuczek et al. on p. 291 ff., the recently synthesized complex $[\{(\eta^5\text{-C}_5\text{Me}_5)\text{Ti}(\eta^6\text{-C}_5\text{H}_4\text{C}(p\text{-tolyl})_2)\}_2(\mu\text{-}\eta^1\text{:}\eta^1\text{-N}_2)]$ (**1**; R. Beckhaus et al., *Eur. J. Inorg. Chem.* **2005**, 1003) is compared to the dinuclear (dinitrogen)titanium complex $[\{(\text{Me}_3\text{Si})_2\text{NTiCl}(\text{TMEDA})\}_2(\mu\text{-}\eta^1\text{:}\eta^1\text{-N}_2)]$ (**3**; R. Duchateau et al., *J. Am. Chem. Soc.* **1991**, 113, 8986). With respect to **3** [$\nu(\text{NN}) = 1284\text{ cm}^{-1}$], **1** exhibits an unusually high N–N stretching frequency [$\nu(\text{NN}) = 1749\text{ cm}^{-1}$]. These spectroscopic differences reflect different degrees of dinitrogen activation, which are explained by the fact that only one of the N_2 π^* -orbitals is occupied in **1**, whereas both π^* -orbitals are occupied in **3**.



MICROREVIEW

Contents

273 L. M. Berreau*

Bioinorganic Chemistry of Group 12 Complexes
 Supported by Tetradentate Tripodal Ligands
 Having Internal Hydrogen-Bond Donors

Keywords: Zinc / Hydrogen bonding / Hydrolysis /
 Cadmium / Group 12

